





The information carried by the <u>DNA bases</u> translates into proteins. The DNA molecule is copied into a different type of nucleic acid—the <u>RNA</u> or ribonucleic acid. The RNA moves to the 'ribosome', an organelle in charge of making proteins. Every set of three bases in the RNA determines which <u>amino acid</u> is added to the protein molecule in progress. The RNA chain passes through the ribosome until the protein is complete.

This process has been called 'the central dogma', as it is the basis of all biological life. Amino acids are bi-functional organic compounds that contain a basic amino group $(-NH_2)$ and an acidic carboxyl group (-COOH). In proteins, 20 different amino acids are commonly found, varying in function and property according to the nature of the R-group. For instance, in alanine, the R-group is $(-CH_3)$, whereas, in cysteine, it is $(-CH_2-SH)$.



In proteins, amino acids are joined together in chains by <u>peptide</u> (amide) bonds that form the molecule's backbone. A peptide bond is formed between a basic amino group $(-NH_2)$ on one amino acid and an acidic carboxyl group (-COOH) on another. The general formula of an amino acid is $H_2 N - CHR - COOH$. The R group can be anything from an atom to a complex molecule. The term 'polypeptide' simply refers to a long chain of amino acids.

Once the protein chain has been made, it must fold up properly before it can do its job. The structure and folding of each protein is specific. How the amino acid sequence causes the protein to fold is not yet completely understood. A 'protein' can be made up of one or more separate polypeptides; it can be made of sheets (amino acid chains lining up together) that contain spiral structures. Clearly, the properties of polypeptides and proteins depend on their amino acid composition.



The secondary structure is the result of local hydrogen bonds being created along the polypeptide backbone. This gives the protein strength and flexibility. Common structures found are:

- *Alpha-helix*, caused by hydrogen-bonding within the polypeptide chain, for example, muscle proteins.
- *Beta-pleated sheet*, caused by hydrogen-bonding between adjacent polypeptide chains, for example, silk fibroin.



The tertiary structure results from interactions between the R-groups in a polypeptide, such as non-covalent bonds (hydrogen bonds, ionic bonds, hydrophobic interactions) and weak, covalent bonds (disulphide bonds between cysteine residues).



Sometimes a single polypeptide is sufficient for the protein to be active; we then talk of a protein that acts as a monomer. Often, however, two or more polypeptides need to interact to allow a protein to perform its particular function. If this is the case, we talk of a dimer; and so on through trimers, etc.

The quaternary structure results from interactions between two or more polypeptide chains to form dimers, trimers, tetramers, etc. They are held together by hydrogen bonds, ionic bonds and, less commonly, hydrophobic interfaces and inter-chain disulphide bonds.



The three-dimensional structure that proteins have is a direct result of interactions with its internal environment. As a consequence, knowing how proteins are structured tells us a lot about how they perform their tasks in the cell.

For instance, in aqueous environments, the hydrophobic R-groups are positioned towards the protein's interior. Changes in temperature or pH can interfere with the non-covalent bonding, causing disruption in the three-dimensional structure and loss of activity. This process is called <u>'denaturation'</u>. Denatured proteins can also clump together to become insoluble in a process called coagulation.

The diversity of protein function, as facilitated by the complexities of protein structure, include (with examples):

- Structural (collagen, muscle fibres)
- Storage (wheat gliadins, barley hordeins)
- Enzymes (hydrolases, transferases, isomerases, polymerases, ligases)
- *Transport* (oxygen transfer with haemoglobin)
- *Messengers* (insulin and certain other hormones)
- Antibodies (proteins that bind to specific foreign particles)
- *Regulation* (proteins involved in regulating DNA synthesis)

<u>Polymorphisms</u> are generated by changes in the amino acids that may result in changes in the primary structure of the enzyme.

Basic reference

Griffiths, A.J.F., J.H. Miller, D.T. Suzuki, R.C. Lewontin and W.M. Gelbart. 1996. The nature of the gene. Pp. 345-358 *in* An Introduction to Genetic Analysis (6th edn.). W.H. Freeman and Co., NY.

